Changes in Phenol Level and Antioxidant Activity of Cocoa Beans

During Fermentation and Roasting

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Received: February 2, 2018	Accepted: March 27, 2018	Online Published: May 20, 2018	
doi:10.5539/jfr.v7n4p23	URL: https://doi.org/10.5539/jfr.v7n4p23		

Abstract

Flavonoid, polyphenols, especially catechin and epicatechin, are major components in cocoa products, which are known for antioxidant properties. Cocoa bean requires fermentation process in order to obtain good taste. During the fermentation process, polyphenols are oxidized by polyphenol oxidase to form quinone and 2-quinon. The purpose of the research is to determine the total level of polyphenol and antioxidant activity of fermented and roasted cocoa beans. The experiment was using Completely Randomized Design (CRD) with fermentation treatment that includes: Without fermentation, three-day fermentation and five-day fermentation. Observation parameters include: Polyphenol level and antioxidant activity with DPPH method. The results showed that the total phenol level of cocoa bean changed during fermentation and after roasting. The highest phenol level was found in cocoa beans without fermentation, and there is a decrease of phenol level to 98% after fermented bean showed a very strongly active antioxidant activity with IC₅₀ of 7.848 ppm, whereas three-day fermentation showed a strong antioxidant activity with IC₅₀ of 35.961 ppm, and five-day fermentation is moderately active with IC₅₀ of 55.976 ppm.

Keywords: cocoa beans, fermentation, poliphenol, antioxidant activity, DPPH

1. Introduction

One of plantation crops that the product is widely used in industrial world is cocoa bean (*Theobroma cacao* L.), because it can be processed into various products with high economic value. Nowadays cocoa products have a lot of attention because it contains a high amount of antioxidant, they even have greater potential than tea and some fruits products that have been known as a source of natural antioxidants(Wilkinson, 1999).

Cocoa contains various polyphenolic compounds stored in the pigment cells of cotyledons (Counet et al., 2004, Wollgast, 2004). The amount of polyphenol in cocoa beans reaches about 6-8% (Zumbe, 1998), it can even reach 10% of the dry weight (Rusconi; Conti, 2010). In addition, cocoa beans contain phenolic compounds such as: catechin, epicatechin, proanthocyanidin, phenolic acid, tannins and other flavonoids. Of this amount, it turns out that flavanol is found in highest quantities in cocoa beans when compared to other types of flavonoids. Cocoa beans have potential as a natural antioxidant ingredients, which we all know that antioxidant could modulate modulate the immune system, prevention of coronary heart, cancer, and as antibacterial (Othman et al. 2007, Weisburger, 2001, Keen et al. 2005., Marsaban, 2007, Sartini et al. 2011). Thus a high level of antioxidant activity makes cocoa could be developed into a healthy product.

Cocoa beans still contain polyphenol although the concentration of polyphenol is decreasing during fermentation and drying. This polyphenol causes bitter and astringent tastes. During fermentation the total amount of phenolic compounds decreased as much as 70%. The sharp decline can be observed between the second and third day, when the seed dies. It is caused when the seeds begin to enter the fermentation of 24-28 hours (dead seeds), polyphenols diffuses from cells then undergo polymerization, oxidation and interactions with proteins.

The next stage of the process after drying, cocoa beans are roasted in 120-140°C for about 45 minutes. Roasting may damage polyphenols and this process could reduce the bitter and astringent tastes in cocoa beans. In such

conditions polyphenolic compounds changed much because of their prolonged exposure to oxygen at relatively high temperatures. The heating process performed causes structural changes of stereoisomers flavanol that may result in changes in antioxidant activity. The benefit of antioxidant product is determined by the level of antioxidant activity.

Measuring the antioxidant activity of intermediate products is needed to determine how much antioxidant potency before and after processed into foods. The method of measurements of the antioxidant activity will detect the different characteristics of the antioxidants in the sample.

Results of a research conducted by Redovnikovic et al. (2009) used two different methods of measuring antioxidants, that is, ORAC (Oxygen Radical Antioxidant Capacity) and DPPH (1,1'-diphenyl-2-picrylhydrazyl), showed that the total polyphenol content in cocca positively associated with antioxidant activity in both methods.

In this research, the measurement of antioxidant activity will be performed with DPPH (2,2-diphenyl-1-picrylhydrazyl). The advantages of the DPPH test method is that it is already widely used in the world and is easy to implement because of the radical compound used is relatively stable compared to other methods (Hartanto, 2012). The purpose of the research is to determine the levels of polyphenols and antioxidant activity of cocoa beans after fermentation and roasting with DPPH (2,2-diphenyl-1-picrylhydrazyl) method.

2. Experimental Methods

The ripe cocoa pods are split, then the seeds are removed, next fermented in wooden box. Fermentation is performed with: unfermented cocoa beans, three-day and five-day fermented seeds,. When the fermentation is complete, the cocoa beans are dried until the water content reaches 6-7%.

Observation Parameter:

2.1 Determination of Polyphenols (Lee et al. 2003)

Making of Standard Solution of Tanat Acid. Standard solution with 100 ppm is made by weighing 10 mg of tannic acid, dissolved with distilled water up to 100 mL. After that, a series of concentration is made, which are 5, 10, 15, 20, 25 ppm. Next is determination of Maximum Wavelength (λ max). It is performed by measuring the solution of 10 ppm tannic acid in the wavelength range of 600-800 nm. The measurement showed that the maximum wavelength is 769.87 nm.

• Measurement of Tanat Acid Standard Solution

Tannic acid solution with a concentration of 5, 10, 15, 20, 25 ppm each is pipetted 1 mL and added with 0.5 mL of Folin, and mixed with vortex method for 2 minutes, then added 2 mL of 7% natrium carbonate then mixed with vortex method and incubated for 30 minutes. Absorbance is measured at a wavelength of 769.87 nm.

- Making of Sample Solution of Cocoa Beans Ethanol Extract This step is done by weighing 10 mg of ethanol extract of cocoa beans that is dissolved in distilled water up to 10 mL, pipettedas much as 1.25 mL, 5 mL inserted into the flask until the concentration reaches 250 ppm.
- Determination of total flavonoid of cocoa beans ethanol extract. 1 mL ethanol extract of cocoa beans is pipetted and added by 0.5 mL of Folin, and mixed with vortex method for 2 minutes, then added 2 mL of 7% Sodium carbonate then mixed with vortex method and incubated for 30 minutes. Absorbance is measured at 769.87 nm wavelength. Replication is performed 3 times.

2.2 Testing of Antioxidant Activity is Performed with DPPH Method as Practiced Modified (Othman et al. 2007)

Preparation of DPPH

- DPPH solution of 0.4 mm is made by weighing 15.7 mg of DPPH powder then diluted with 100 ml of ethanol (p.a) in a volumetric flask. The measurement of maximum wavelength is performed by pipetting 1 mL DPPH solution and added by 3.5 ml of (p.a) ethanol then incubated for 30 min at 30 °C and the absorbance is measured at 450-800 nm wavelength.
- Preparation and measurement of antioxidant potency of quercetin comparator sample.
- Standard solution of 1000 ppm is made by weighing 10 mg quercetin, diluted with ethanol (p.a) up to 10 mL. Then 100 ppm of standard solution is made by pipetting 0.5 mL of 1000 pm standard solution until it reached 5 mL. After that, a series of concentration (4, 6, 8, 10, 12 ppm) is made,. Testing is performed with pipetting 1 mL of standard solution from various concentrations and then each is added with 0.4 mm of DPPH then added 2.5 ml of ethanol (p.a). Solution is incubated at 37 °C for 30 minutesand measured at 515.97 nm wavelength.

- Preparation and measurement of antioxidant potency of cocoa beans ethanol extract

1000 ppm of stock solution is made by weighing 25 mg of the ethanol extract of cocoa beans and dissolved with ethanol (p.a) up to 25 ml in a flask. Then various concentrations of 500, 600, 700, 800, 900 ppm is made. Testing is performed by pipetting 1 mL of various concentrations and then each is added DPPH 0.4 mm then added 2.5 mL of ethanol (p.a). Solution, incubated at 37 $^{\circ}$ C for 30 minutes and measured at a wavelength of 515.97 nm. The amount of the antioxidant strength is calculated using the formula (Othman et al. 2007):

% inhibition = $(\frac{blank \ Absorbance - sample \ Absorbance}{blank \ Absorbance})x100 \%$

 IC_{50} is calculated by plotting scavenging activity graphs with concentrations of the extract, which is defined as the total antioxidant needed to lower the concentration of DPPH to 50%.

3. Results and Discussion

3.1 Results

a. Phenol Content in Cocoa Beans After Fermented

During fermentation, there would occur decomposition of polyphenol, protein and sugar compounds in cocoa beans by the enzymes that would produce aroma candidate compounds, taste improvement and color change. Determination of total phenol content of dry beans was performed using the Folin-Ciocalteu method. The use of tannic acid which is a component of the phenol as a standard would predict phenol level in cocoa beans. Phenol content of fermented dry cocoa beans can be seen in Figure 1.

ANOVA testing ($\alpha = 0.01$) showed that the fermentation time of cocoa beans affects the phenol level significantly. Unfermented cocoa beans had the highest phenol content, while the 5-day fermentation of cocoa beans contained the lowest, but not significantly different from the phenol level of 3-day fermentation of cocoa beans. Phenol level decreased by approximately 50% after fermentation.

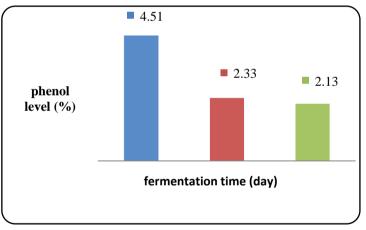


Figure 1. Phenol level of dried cocoa beans after fermentation

Figure 1 shows the total phenol content of unfermented cocoa beans is 5,00% and decreased by 50% in 3 days of fermentation (2.33%) and 5 days to 2.13%.

b. Total Phenol of Cocoa Beans After Roasting

Fermented cocoa beans were dried, then roasted at temperature of 120°C for 40 minutes.ANOVA test showed that roasting significantly affected the phenol content of cocoa beans. Roasting led to decrease of phenol level in cocoa beans. This is proved from the test which shown that the unfermented cocoa beans had the highest phenol level (0.534%), while the five days fermented cocoa beans had the lowest phenol level (0.195%).

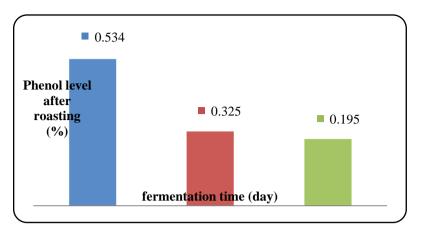


Figure 2. Total phenol content after roasting

Figure 2 shows the phenol level declined very sharply after roasting. The magnitude of the decline rate of phenol level reached an average of 98%, both fermented and unfermented beans.

c. Testing of Antioxidant Activity

Measurement of cocoa beans extract absorbance carried out at a wavelength of 517 nm, Absorbance resulted was then converted intoInhibition Concentration (IC_{50}). The measurement results can be seen in Table 1.

Fermentatio (day)	Concentration (ppm)				IC ₅₀	
	10	20	30	40	50	
0	44.780	67.931	89.699	94.118	95.874	7.958a
3	34.324	39.944	44.517	56.155	58.111	36.926b
5	30.288	36.351	38.635	43.890	55.986	56.986c
Quercetin/concentration	0.5	1	2	3	4	
	31.942	38.653	43.235	49.764	55.257	2.996

Table 1. The antioxidant activity of cocoa bean and quercetin extract using DPPH

Table 1 shows the fermentation time significantly affect the IC_{50} value of cocoa beans. Unfermented cocoa beans have IC_{50} value of 7.958, whereas the fermented cocoa beans obtained higher values. The IC_{50} highest value was shown in the 5-days fermented cocoa beans.

3.2 Discussion

Polyphenol

The content of polyphenol and antioxidant activity in the unfermented cocoa beans is higher than in the fermented cocoa beans (Prayoga, Murwani, & Anwar, 2013), because oxidation, polymerization, and protein binding process could reduce the polyphenol level in cocoa beans during fermentation (Nazaruddin et al., 2006). In general, the fermentation process of cocoa beans may take between 5 to 7 days, and there are two processes, namely external and internal fermentation. External fermentation is the process of sugar catabolism by microorganisms, whereas internal fermentation includes biochemical processes into the seed cotyledons. During the fermentation process, polyphenol compounds disperse out of their storage cells and then undergo oxidation that forms high molecular compounds (Bernaert et al., 2012). As shown in Figure 1, phenol content of cocoa beans changes during the fermentation process. The highest value of total phenol content was found in unfermented cocoa beans with the value of 4.51% and a phenol content decreased after fermented for 3 days with values of 2.33% and to 2.13% after 5 days of fermentation. The decrease in phenol levels during fermentation was suspected caused by the hydrolysis and oxidation of phenolic compounds during the fermentation process. The decline is due to the oxidation of polyphenols by the activity of the oxidase polyphenol enzyme (PPO). The oxidase polyphenol enzyme is an enzyme that is essential in the process of oxidation of polyphenol compounds in cocoa. The enzyme started working during the fermentation process and continued until the early stages of the drying process(Abdulamir et al. 2012).

All the processes that occur caused a significant reduction in polyphenol content of the seed. The amount of

cocoa polyphenols, especially (-) - epicatechin was reported decreased by 90% or more during the fermentation and drying (Kim & Keeney, 1984; Pettipher, 1986; Wollgast and Anklam, 2000). The content of polyphenols decreased sharply in the second and third day when the seed died. When the seeds began entering the fermentation of 24-28 hours (dead seeds), polyphenols diffused from cells that subsequently underwent polymerization and oxidation. Oxidase polyphenol achieved optimal activity at pH 6 and temperature of $35.5 \,^{\circ}$ C (Lopez and Dimick, 1995; Misnawi and Selamat, 2003). The oxidase polyphenol enzyme activity would decrease rapidly towards the end of fermentation, it might be caused due to an increase in temperature, and oxidase polyphenol would not be active when the beans are dried (Lopez and Dimick, 1995; Afoakwa et al. 2012).

After drying, the cocoa beans are roasted in 120-140^oC for approximately 40 minutes. Figure 2 shows that the phenol level in both fermented and unfermented cocoa beans could decrease up to 98% after roasting. During roasting polyphenols was damaged so that bitter and astringent taste are reduce. In such conditions polyphenolic compounds might change much since long exposed to oxygen at relatively high temperature. The heating process performed caused structural changes of stereoisomer flavonoids that might result in changes in antioxidant activity. Till nowadays, roasting of cocoa beans are still performed conventionally with the rate of transferred heat is limited to three standard ways, which are conduction, convection and radiation. These method led to slow heating process. In such circumstance, polyphenolic compounds changed much since the prolonged contamination to oxygen at relatively high temperature, so it affects the antioxidant activity (Supriyanto et al. 2006)

Antioxidant Activity

Antioxidant testing of phenolic compounds can be performed by using DPPH radical. Essential structure required for anti-radical activity was the presence of free electrons species. Compound having a hydrogen atom that can be released easily can react with DPPH radical. Most antioxidant compounds were known to possess phenol group that can easily donate hydrogen radical to the DPPH radical (Makhmoor, 2005; Shetty and Wahqvist, 2004). The antioxidant activity is the ability of a compound or extract to inhibit the oxidation reaction which can be expressed as inhibition percentage. Parameters used to indicate the antioxidant activity is efficient concentration (EC_{50}) or Inhibition Concentration (IC_{50}) which is an antioxidant concentration that can cause 50% of DPPH lost its radical character. As stated by Brand-Williams et al. (1995), substances that have a high antioxidant activity, would have a lower EC_{50} or IC_{50} .

The antioxidant compound will react with the DPPH radical through a hydrogen atom donation mechanism and cause color discoloration of DPPH from purple to yellow as measured at a wavelength of 517 nm (Hanani et al., 2005, Rahayu et al. 2009). The incubation time for 30 minutes at 37° C was required to ensure the DPPH radical works. Inhibition Concentration (IC₅₀) is the concentration of an antioxidant substance that can cause 50% DPPH to lose the radical character or in other words, it is a concentration of an antioxidant substance that gives 50% inhibition (Suratmo, 2008).

The IC₅₀ values of cocoa beans extract from the three treatments are: unfermented has 7.848 ppm classified as very strongly active, 3-days fermentation amounted to 35.961 ppm classified as strongly active, and 5-days fermentation has 55.976 ppm that is classified as moderately active. The lower the IC₅₀ value of a sample, the higher the antioxidant activity. Deterrence activities of DPPH free radicals of the cocoa bean extract of the three treatments compared to quercetin as standard showed that extracts of cocoa beans were smaller than the quarsetin comparator which is 2.896 ppm, and categorized as "very strongly active". Bioactive compounds can be classified based on the IC₅₀ value. Value of IC₅₀> 250 ppm is classified as "inactive"; 100-250 ppm as "weakly active"; 50-100 ppm as "moderately active"; 10-50 ppm as "strongly active" and <10 ppm as "very strongly active" (Phongpaichit et al. 2013).

4. Conclusion

The fermentation, drying, and roasting of cocoa beans interacted with one another in influencing the level of polyphenols and antioxidant activity. The process of fermentation and roasting of cocoa beans can decrease the polyphenol content by 98%. In addition, fermentation and roasting greatly affect the antioxidant activity of cocoa beans, compared to unfermented cocoa beans. Beside that, fermentation and roasting highly affect the antioxidant activity of cocoa beans. Cocoa beans that were not fermented showed a very strongly active antioxidant activity with IC_{50} 7.848 ppm, while the 3-day fermentation showed a strong antioxidant activity with IC_{50} 35.961 ppm and 5-day fermentation showed moderately active with IC_{50} 55.976 ppm.

Acknowledgments

The authors would like to thank the Ministry of Research and Technology and Higher Education who have funded the research in the Competitive Grant scheme so that this research can be done.

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