Quality Evaluation of Beef Preserved With Food Grade Organic Acids at Room Temperature

Eniolorunda O. O.¹, Apata E. S.¹, Ogunlesi O. E.¹, & Okubanjo A. O.²

¹ Meat Science Laboratory, Department of Animal Production Olabisi Onabanjo University, Ogun State, Nigeria
² Department of Agriculture and Industrial Technology, Babcock University, Ogun State, Nigeria

Correspondence: Apata E. S., Meat Science Laboratory, Department of Animal Production Olabisi Onabanjo University, Ogun State, Nigeria. E-mail: ebuoluapata2008@yahoo.com

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Abstract

The objective of this study was to investigate the effects of three common food grade organic acids – citric, acetic and ascorbic on quality properties of fresh beef preserved for 14 days. 1 kg of fresh beef (thigh muscle) of White Fulani cow was purchased at Ayetoro market in Yewa North local government Area of Ogun State and was divided into 4 equal parts of 250 g per treatment replicated three times. The acids were purchased at Federal Institute of Industrial Research Oshodi (FIIRO) Lagos. 5% each of the organic acid was prepared and constituted an experimental treatment, freezing was used as control. Thus: T1 = Freezing (control), T2 = Citric acid, T3 = Acetic acid, T4 = Ascorbic acid. 10ml of each organic acid solution was injected into 250 g fresh beef with a needle and syringe and immersed in the same solution in covered plastic containers, stored at room temperature (27 ºC).

The results showed that most of the physicochemical properties of the preserved beef were better (P < 0.05) in treatment 3, also. Lipid oxidation and microbial values were lower (P < 0.05) in the same treatment. However, acceptability of beef in treatment3 was lower (P < 0.05) because colour and fl avour scores beef were lower (P < 0.05). It was suggested therefore, that lower concentrations of acetic acid be tested in a separate study to ascertain concentration level that will confer higher colour flavour and acceptability scores on beef since acetic acid favoured almost all tested properties and of preserved beef in this study.

Keywords: beef, food grade, organic acids, quality, room temperature

1. Introduction

Beef when fresh is high in nutritional value in that it is rich in vitamins and minerals and provides an important source of high quality protein (Mohamed et al., 2008). It has unique biological and chemical attributes and its nutrient composition predisposes it to deterioration due to microbial growth and rancidity development, hence its shortened shelflife (Houben et al., 2000). The shelf-life represents the useful storage time of meat or meat product, beyond this period, changes in smell, colour, taste and texture make them unacceptable. It is affected by several factors such as temperature, pH, oxygen, pressure, light and oxidation (Shahidi, 1994). Different types of pathogenic microorganisms may be introduced into and on the surface of fresh beef during slaughtering and processing, which cause rapid spoilage, and great loss of valuable protein, therefore, interventional procedures have been used to decontaminate the meat (Stopforth & Sofos, 2006). One of such interventional procedures is the use of organic acids. Since fresh beef spoilage results from the activity of mixed populations of microorganisms, food grade organic acids can be used as a single or in combination for effective reduction of spoilage due to pathogenic microorganisms – (Lebert et al., 2005). Solutions of organic acids (1-5%) such as lactic, acetic, citric, ascorbic, propionic fumaric and tartaric acids are the most frequently used chemical interventions for both beef and lamb dressing (Acuff, 2005). Acetic acid is a mono carboxylic acid with a pungent odour and taste known as vinegar which has antimicrobial capabilities due to its ability to lower the pH and cause instability of bacterial cell membrane (Ransom et al., 2003). Citric acid is a 2-hydoxy-1,2,3-propane tricarboxylic acid (white powder) extracted from juice of acidic fruits capable of inhibiting the growth of bacteria, yeast and moulds (Dubal et al., 2004), while L-ascorbic acid is slightly soluble in water and has antimicrobial capabilities like other organic acids (Wicklund et al., 2005). These three organic acids (acetic, citric and L-ascorbic) could be used in preserving fresh beef or any other types of fresh meat in rural areas of...
developing countries like Nigeria where electricity supply is still epileptic since they are common and can prolong the shelf life of fresh meat up to eleven days in refrigerated temperature (Dubal et al., 2004). There is inadequate information in literature concerning the consequences of preserving fresh beef meat directly in organic acids at room temperature. Therefore, this study was conducted to evaluate the potentials and effects of the three food grade organic acids on quality characteristics of fresh beef.

2. Materials and Methods

2.1 Location of Study

This study was conducted in the Meat Science Laboratory of the Department of Animal Production, Olabisi Onabanjo University, Yewa Campus, Ayetoro, Ogun State.

2.2 Meat Samples

One kilogramme of fresh beef (*Longissimus dorsi*) of White Fulani cow was purchased and used for this study. It was divided into 4 equal parts of 250 g per treatment and was replicated three times.

2.3 Organic Acids

The dietary organic acids tested were citric, acetic and ascorbic acids. They were purchased from Federal Institutes of Industrial Research, Oshodi, (FIIRO) in Lagos. 5% of each organic acid was prepared with distilled water. Each of the organic acid constituted an experimental treatment, while cold preservation by freezing was used as control treatment as arranged below:

- **T1 (control)** = Freezing preservation
- **T2** = Citric Acid
- **T3** = Acetic Acid
- **T4** = Ascorbic Acid

2.4 Preservation of Meat

10ml of each tested solution of 5% organic acid was injected with the aid of a needle and symige into 250 g fresh beef meat sample and was immersed in the same solution of each organic acid in a plastic container with a lid and stored at room temperature (27 °C) in a netted box for 14 days (2 weeks) after which the following meat parameters were determined.

2.5 Raw Meat Colour

This was determined with visual method following the procedures of AMSA (1991). Meat samples from each treatment were displayed in a tray and a 10-member panel was used to score the meat based on colour intensity (redness) and homogeneity of the meat samples using a scale ranging from 1-8 with higher scores representing a more attractive and homogenous red colour.

2.6 Cooking Loss and Thermal Shortening

Cooking loss was measured by removing approximately 10 g and 6 cm long meat sample from each treatment, wrapped in air tight polythene bags with a thermometer (110 °C) inserted in the meat and cooked in water in a pre-heated cooking pot for 20 min. on and adjustable Pifco Japan Electric “hot” plate Model No. ECP 2002 until the geometric centre of the meat samples was heated to 72 °C (Malgorzata et al., 2005). Meat samples were removed from the pot and cooled to room temperature (27 °C). They were reweighed and the difference in weight recorded as percentage cooking loss as follows:

\[
\text{Cooking loss} = \frac{\text{Initial wt. of meat} - \text{Final wt. of meat}}{\text{Initial Wt. of meat}} \times 100
\]

Thermal shortening of the meat samples was measured with the same meat samples used to measure cooking loss. The lengths of meat samples were remeasured after cooking and cooling, the difference in length was expressed as thermal shortening following the modified method of Malgorzata et al. (2005).

Thus:

\[
\text{Thermal shortening} = \frac{\text{Initial length of meat} - \text{Final length of meat}}{\text{Initial length of meat}} \times 100
\]
2.7 Percentage Cooking Yield
This was obtained by subtracting the value of percentage cooking loss from 100% and the remainder recorded as the percentage cooking yield according to Omojola (2008).
Thus: Cooking yield = 100% - % cooking loss

2.8 Drip Loss
This was determined following the procedures of Insausti et al. (2001). Weight of an empty polythene bag was taken (Wp) Meat sample (10 g) was weighed and put into the bag (Wp + M) and stored in a refrigerator at 4 ºC for 48 h. The meat sample was removed from the refrigeration and the weight of the bag plus the juice drained by the meat sample were measured (Wp + j). drip loss was expressed as percentage of the initial weight of the meat sample, thus:

\[ \text{Drip loss} = \frac{(Wp + j) - (Wp)}{(Wp + m) - (Wp)} \times 100 \]

2.9 Water Holding Capacity (WHC)
An approximately 1g of meat sample from each treatment was placed between two 9 cm Whatman No 1 filter papers (Model C, Caver Inc. Wabash, USA). The meat sample was pressed between two 10.2 × 10.2 cm² plexiglasses at about 35.2 kg/cm² absolute pressure for 1 minute with a vice (Suzuki et al., 1991). The meat sample was removed and oven dried between 100-105 ºC for 24 h to determine the moisture content of the meat sample. The amount of water released from the meat sample was measured indirectly by measuring the area of filter paper welted relative to the area of pressed meat sample. Thus:

\[ \text{WHC} = \frac{100 - (Aw - Am)}{Wm \times Mc} \times 9.47 \times 100 \]

Where:
Aw = Area of water released from meat sample (cm²)
Am = Area of meat sample (cm²)
Wm = Weight of meat sample (g)
Mc = Moisture content of meat sample (%)
9.47 = A constant factor.

2.10 Shear Force
Weighed meat samples from each treatment (10 g) was wrapped in polythene bags and cooked in a pre-heated cooking pot for 20 min on an adjustable Pifco Japan Electric hot plate Model NECP 202 to an internal temperature of 72 ºC. They were removed and cooled to room temperature (27 ºC) for 10min, reweighed, bagged and chilled at 4 ºC for 18 h. They were equilibrated to room temperature and 1.25cm diameter cores parallel to muscle fibre orientation were removed with a coring device (Qiaofen & Da-Wen, 2005). The meat samples were sheared at three locations with WarnerBratzler V-notch blade shearing instrument according to Honikel (1998) and average value of the three shearing was taken.

2.11 Proximate Composition
This was determined following the procedures of AOAC (2000). Moisture content of meat samples was obtained by drying 2 g of meat in an oven at 100-105 ºC for 24h until a constant weight was reached. Crude protein was determined by using Kjedahl method which comprised, digestion, distillation and titration of the distillate. Crude protein value was obtained by converting nitrogen (N%) content with a constant (6.25) crude protein was thus obtained as (6.25 × N%). Fat was determined with soxhlet extraction method using petroleum ether. Ash content of the meat sample was determined by igniting 2g of it in a Muffle furnace set between 550 and 600 ºC for 4 h until the ash was formed and was weighed.

2.12 pH of Meat
This was taken after the preservation period of 14 days. 10 g of meat sample from each treatment was homogenized for 5 min with 90 ml distilled water with a blender (plate 5 mm) model 242, Nakai, Japan. The
meat pH was measured with a portable pH meter Model H184 Micro Computer, Havanna Instruments, Romania (Marchiori & de Felicio, 2003).

2.13 Lipid Oxidation

Lipid oxidation of meat sample from each treatment was determined with TBA and mPV methods; TBA – This was determined using the method described by Pensel (1990). 5.0 g of coarsely ground unrendered fat of meat sample from each treatment was placed in a polyethylene bag. An additional empty polyethylene bag was prepared as a blank. 50 ml of a cold (2±2 °C) 20% trichloroacetic acid (TCA) and 1.6% m-phosphoric acid mixture was immediately added to each meat in polyethylene bags and ground in a blender (Plate 5 mm) (Model 242 Nakai Japan) for 2 min. 50 ml of cold (2±2 °C) distilled water was added to each bag and blended for an additional 30 sec. The slurry was filtered through Whatman No. 1 filter paper to remove the debris. 5.0 ml of the filtered slurry was added to 5.0 ml of freshly prepared 0.02 M 2-thiobarbituric acid and mixed for 5 sec. The samples were subsequently stored in the dark at room temperature (27 °C for 15 hrs for the colour to develop. The colour was measured using a Gilford Response UV-VIS spectrophotometer (Ciba corning Diagnostic CO, Oberlin Ohio) at a wavelength of 530 nm.

MPV – This was determined following the modified procedures of AOAC (2000). 50 g of unrendered fat of meat sample from each treatment was ground in a blender (Plate 5 mm) (Model 242, Nakai, Japan) for 25 sec and extracted with 30 ml of ice cold (3:2 V/V) acetic acid: chloroform. The extraction was vigorously swirled to distribute the samples and reagents evenly. 0.5 ml of saturated potassium iodide (K1) was added and mixed thoroughly. 30 ml of distilled water was added subsequently and the solution was mixed thoroughly. The mixture was allowed to stand for 5-10 mins, at room temperature (27 °C). The mixture was titrated with 0.01 M sodium thiosulphate (Na2S2O3) gradually with vigorous shaking. 0.5 ml of starch indicator (1% starch + 0.3% chloroform) was added during titration. The sample was vigorously swirled and was allowed to stand for an additional 10 mins. The end point was established when the colour of the upper aqueous layer disappeared. The modified peroxide value (mPV) of samples was calculated with the formula:

\[
\text{mPV} = \frac{S \times N \times 1000}{W}
\]

Where:
- \( mPV \) = modified peroxide value (mEq)
- \( S \) = ml of Na2S2O3 (0.5)
- \( N \) = Normality of Na2S2O3 (0.01)
- \( W \) = Weight of sample (g).

2.14 Microbiology Evaluation of Meat

10 g of meat sample was removed from each treatment and blended with 90 ml of 0.1% (W/V) peptone water for 60 sec. Dilutions were made in 0.1% peptone water and 1 ml of undiluted homogenate of each sample was spread on duplicate petriplates. Bacterial counts were obtained from plates bearing colonies as follows: Aerobic plate count – on Plate count agar (DIFCO, USA) incubated at 32 °C for 48 h to isolate staphylococcus; Enterobacteriaceae (Coliform) on violet Red Bile Glucose agar (DIFCO, USA) overlaid with the same medium and incubated at 37 °C for 24 h to isolate Enterobacteria spp. and Lactic Acid Bacteria (LAB) on Lactobacilli MRS Broth, Bacto agar and glacial acetic acid (pancreae) and incubated at 32 °C for 48 h to isolate lactobacillus spp. Counts were made and expressed in cfu/g of samples following the procedures of ICMSF (1986), APHA (1992) and AOAC (2000).

2.15 Sensory Evaluation of Meat

A 10-member semi-trained taste panel was used following the procedures AMSA (1995). The panellists were instructed on how to fill the sensory evaluation form. They were provided unsalted biscuits and water for use in between treatments meat samples. Meat samples were cooled after cooking for 20 min in labeled polythene bags with a thermometer (110 °C) inserted into the meat to an internal temperature of 72 °C using moist cooking. They were presented sequentially to the panellists on a clean saucer and meat sample from each treatment was evaluated independently. The panellists rated the meat samples on a 9-point hedonic scale on which 1 = dislike extremely and 9 = like extremely for colour, odour, flavour, tenderness, juiciness, ropiness, cohesiveness and overall acceptability.
2.16 Experimental Design and Statistical Analysis

Completely randomized design was used for study. Data collected were subjected to analysis of variance (ANOVA) at (p < 0.05) using (SAS, 2002) and significant different among the means were separated with the aid of Duncan multiple range test of the same system.

3. Results and Discussion

All physical attributes except raw meat colour, Water Holding Capacity (WHC) and shear force were better (P < 0.05) in treatment 1 than in other treatments (Table 1). Meat colour is enhanced by contact with oxygen as myoglobin is oxidized into oxymyoglobin (glooming). Meat sample in treatment 1 might have been in contact with oxygen in the freezer throughout the preservation period, whereas meat samples in treatments 2, 3 and 4 might not due to the fact that the plastic container in which the meat samples were preserved with organic acids which were completely covered, this might have shielded oxygen penetration into the meat. The pH status of the meat which was higher (P < 0.05) in treatment 1 than in other treatments could be due to high water holding capacity (Abril et al., 2002). Water holding capacity of meat samples was lower (P < 0.05) in treatment 3, probably as a result of low (P < 0.05) pH (Mohamed et al., 2008), most of the water in form of juice in the meat might have been lost into the acidic medium which could have led to higher (P < 0.05) shearforce value due to shrinkage imposed on the meat sample by organic acid (Acetic acid).Since the pH of meat in treatment 1 was high (P < 0.05) it induced higher (P < 0.05) WHC (Miller, 2001). Table 2 shows the results of proximate composition and pH of beef. Moisture content and pH were lower (P < 0.05) in treatment 3, while crude protein and nitrogen free extract (NFE) were higher (P < 0.05) compared with those of other treatments. There were no significant (P > 0.05) effects of organic acids used on fat and ash contents of meat samples across all the treatments in this study. High protein content and NFE in treatment 3 could be attributed to low moisture content and pH which might have prevented protein denaturation and nitrogen degradation by either intrinsic enzymes (catepsins) or microorganisms (Koohmaraie et al., 2005).

Therefore, high pH observed in treatment 1 and 4 could predispose the meat samples to spoilage and shorter shelf-life (Bucchman & Golden, 1994). Lipid oxidation (TBA) was higher (P < 0.05) in treatment 1 than in treatments 2, 3, and 4 while (mPV) was higher (P < 0.05) in treatment 1 followed by treatment 2 and was lower (P < 0.05) in treatments 2 and 3 respectively (Table 3). The high values of TBA and mPV observed in treatments 1 and 4 followed the patterns of WHC and moisture contents of the meat samples. Fat degrading enzymes are more active when moisture and pH are high in meat (Decker & Crum, 1991) therefore, the results of lipid oxidation obtained in this study corroborated that of the previous workers as meat samples in treatments 1 and 4 were prone to oxidative rancidity than meat samples in treatments 2 and 3. Table 4 presented the microbial load of beef meat preserved with organic acids. Treatment 1 had highest (P < 0.05) Aerobic bacteria count (staphylococcus) (6.5 × 10^6) followed by treatment 4 (5.9 × 10^6), while treatment 3 had lowest (P < 0.05) staphylococcus count of 3.5 × 10^6. Treatment 3 had the least (P < 0.05) enterobacteria count (enterobacter spp.) of 3.2 × 10^3 while treatment 1 had the highest (P < 0.05) with 5.4 ×10^3 count followed by treatment 4 with 4.5 ×10^3 count. Moisture contents and pH of meat samples in treatment 1 and 5 were high which might have led to high microbial counts while, those of meat samples in treatments 2 and 3 were low thereby discouraged high population of both staphylococcus and enterobacter spp. in the meat samples. (Young & Fregeding, 1993). Lactic acid bacteria (Lactobacillus spp.) (LAB) count was higher (P < 0.05) in treatment 3, followed by treatment 2, while it was lower (P < 0.05) in treatments 1 and 4. This was possible probably LAB could thrive well in acidic medium than other two species of bacterial and therefore, out grewed rapidly other species in the meat (Lee & Yoon, 2001). They opined that under anaerobic conditions the growth of LAB was favoured against aerobic bacteria thereby extending the shelf-life of the meat. Similar results were observed in this study aerobic bacteria counts were higher in treatments 1 and 4, while those of LAB were lower, but LAB counts were higher in treatments 3 and 2, while those of aerobic bacteria were lower, hence, spoilage of meat samples in these treatments was not as high as observed in TBA and mPV of meat samples in treatments 1 and 4. Cooked meat colour was higher (P < 0.05) in treatment 1 and least (P < 0.05) in treatments 3. The results agreed with (Mikel et al., 1996) who reported that acetic acid deteriorated the surface colour of meat. Treatment 3 gave the meat samples lower (P < 0.05) odour and ropiness scores against treatment 5 with higher (P < 0.05) scores. However, flavour, tenderness, juiciness, cohesiveness and overall acceptability scores were higher (P < 0.05) in treatment 1 while treatment 3 had the least (P < 0.05) scores of these attributes. The lower overall acceptability of meat samples in treatment 3 could be borne out of the fact that it had lower colour and flavour which are the most cherished attributes of meat by consumers.
Table 1. Physical attributes of preserved fresh beef meat as affected by organic acids

<table>
<thead>
<tr>
<th>Treatments</th>
<th>Variable</th>
<th>T1</th>
<th>T2</th>
<th>T3</th>
<th>T4</th>
<th>SEM</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Raw meat colour</td>
<td>6.55a</td>
<td>5.21b</td>
<td>4.00c</td>
<td>5.00b</td>
<td>2.41</td>
</tr>
<tr>
<td></td>
<td>Cooking loss (%)</td>
<td>38.90a</td>
<td>27.90c</td>
<td>20.70d</td>
<td>30.40b</td>
<td>1.69</td>
</tr>
<tr>
<td></td>
<td>Cooking yield (%)</td>
<td>61.10d</td>
<td>72.10b</td>
<td>79.30a</td>
<td>69.60c</td>
<td>1.02</td>
</tr>
<tr>
<td></td>
<td>Thermal shortening (%)</td>
<td>36.80a</td>
<td>25.60c</td>
<td>18.60d</td>
<td>27.20b</td>
<td>1.94</td>
</tr>
<tr>
<td></td>
<td>Drip loss (%)</td>
<td>52.00a</td>
<td>40.00c</td>
<td>30.00d</td>
<td>45.00b</td>
<td>1.28</td>
</tr>
<tr>
<td></td>
<td>Water holding capacity (%)</td>
<td>65.00a</td>
<td>49.10c</td>
<td>43.00d</td>
<td>53.00b</td>
<td>1.35</td>
</tr>
<tr>
<td></td>
<td>Shear force (kg/cm²)</td>
<td>5.26b</td>
<td>6.52a</td>
<td>6.30a</td>
<td>6.25a</td>
<td>2.37</td>
</tr>
</tbody>
</table>

abcd: Means on the same row with different superscripts are statistically significant (P < 0.05).

Table 2. Proximate composition and pH of preserved fresh beef meat as influenced by organic acids

<table>
<thead>
<tr>
<th>Treatments</th>
<th>Variable</th>
<th>T1</th>
<th>T2</th>
<th>T3</th>
<th>T4</th>
<th>SEM</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Moisture (%)</td>
<td>73.00a</td>
<td>72.00b</td>
<td>70.00c</td>
<td>72.60ab</td>
<td>1.63</td>
</tr>
<tr>
<td></td>
<td>Crude Protein (%)</td>
<td>15.30c</td>
<td>16.55b</td>
<td>17.80a</td>
<td>16.40b</td>
<td>2.06</td>
</tr>
<tr>
<td></td>
<td>Either Extract (Fat) (%)</td>
<td>2.70</td>
<td>2.60</td>
<td>2.60</td>
<td>2.70</td>
<td>6.06</td>
</tr>
<tr>
<td></td>
<td>Ash (%)</td>
<td>0.90</td>
<td>0.80</td>
<td>0.80</td>
<td>0.90</td>
<td>0.10</td>
</tr>
<tr>
<td></td>
<td>NFE (%)</td>
<td>7.60b</td>
<td>8.05ab</td>
<td>8.80a</td>
<td>7.40b</td>
<td>4.06</td>
</tr>
<tr>
<td></td>
<td>pH</td>
<td>6.87a</td>
<td>4.76c</td>
<td>4.50c</td>
<td>5.80b</td>
<td>1.20</td>
</tr>
</tbody>
</table>

abcd: Means on the same row with different superscripts are statistically significant (P < 0.05).
NFE = Nitrogen Free Extract.

Table 3. Lipid oxidation of preserved fresh beef meat as influenced by organic acids

<table>
<thead>
<tr>
<th>Treatments</th>
<th>Variable</th>
<th>T1</th>
<th>T2</th>
<th>T3</th>
<th>T4</th>
<th>SEM</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>TBA (ug/g)</td>
<td>0.02a</td>
<td>0.01b</td>
<td>0.01b</td>
<td>0.01b</td>
<td>0.11</td>
</tr>
<tr>
<td></td>
<td>mPV (mEq)</td>
<td>0.30a</td>
<td>0.20c</td>
<td>0.20c</td>
<td>0.21b</td>
<td>0.48</td>
</tr>
</tbody>
</table>

abc: Means on the same row with different superscripts are statistically significant (P < 0.05).
TBA = ThiobarBaturic Acid, MPV = Modified Peroxide Value, TVA = Total Volatile Acids.

Table 4. Microbial load of preserved fresh beef meat as influenced by organic acids

<table>
<thead>
<tr>
<th>Treatments</th>
<th>Variable</th>
<th>T1</th>
<th>T2</th>
<th>T3</th>
<th>T4</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Aerobic Bacteria</td>
<td>6.5×10⁶a</td>
<td>4.7×10⁶b</td>
<td>3.5×10⁶d</td>
<td>5.9×10⁶b</td>
</tr>
<tr>
<td></td>
<td>Enterobacteria</td>
<td>5.4×10⁵a</td>
<td>4.3×10⁵b</td>
<td>3.2×10⁵c</td>
<td>4.5×10⁵b</td>
</tr>
<tr>
<td></td>
<td>Lactic Acid Bacteria</td>
<td>4.2×10⁴c</td>
<td>5.2×10⁴b</td>
<td>6.3×10⁴a</td>
<td>4.9×10⁴b</td>
</tr>
</tbody>
</table>

abc: Means on the same row with different superscripts are statistically significant (P < 0.05).
Table 5. Sensory Properties preserved fresh beef meat as affected by organic acids

<table>
<thead>
<tr>
<th>Treatments</th>
<th>T1</th>
<th>T2</th>
<th>T3</th>
<th>T4</th>
<th>SEM</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cooked meat colour</td>
<td>7.00&lt;sup&gt;a&lt;/sup&gt;</td>
<td>6.00&lt;sup&gt;b&lt;/sup&gt;</td>
<td>4.21&lt;sup&gt;d&lt;/sup&gt;</td>
<td>5.30&lt;sup&gt;c&lt;/sup&gt;</td>
<td>2.37</td>
</tr>
<tr>
<td>Odour</td>
<td>5.00&lt;sup&gt;a&lt;/sup&gt;</td>
<td>3.20&lt;sup&gt;c&lt;/sup&gt;</td>
<td>3.00&lt;sup&gt;c&lt;/sup&gt;</td>
<td>4.10&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.24</td>
</tr>
<tr>
<td>Flavour</td>
<td>6.89&lt;sup&gt;a&lt;/sup&gt;</td>
<td>5.40&lt;sup&gt;b&lt;/sup&gt;</td>
<td>4.21&lt;sup&gt;c&lt;/sup&gt;</td>
<td>4.37&lt;sup&gt;c&lt;/sup&gt;</td>
<td>2.40</td>
</tr>
<tr>
<td>Tenderness</td>
<td>6.80&lt;sup&gt;a&lt;/sup&gt;</td>
<td>5.45&lt;sup&gt;b&lt;/sup&gt;</td>
<td>5.20&lt;sup&gt;b&lt;/sup&gt;</td>
<td>5.60&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.81</td>
</tr>
<tr>
<td>Juiciness</td>
<td>6.11&lt;sup&gt;a&lt;/sup&gt;</td>
<td>4.55&lt;sup&gt;b&lt;/sup&gt;</td>
<td>4.35&lt;sup&gt;b&lt;/sup&gt;</td>
<td>4.59&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.16</td>
</tr>
<tr>
<td>Ropiness</td>
<td>7.40&lt;sup&gt;a&lt;/sup&gt;</td>
<td>5.30&lt;sup&gt;c&lt;/sup&gt;</td>
<td>4.10&lt;sup&gt;d&lt;/sup&gt;</td>
<td>6.20&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.04</td>
</tr>
<tr>
<td>Cohesiveness</td>
<td>7.60&lt;sup&gt;a&lt;/sup&gt;</td>
<td>6.40&lt;sup&gt;b&lt;/sup&gt;</td>
<td>5.20&lt;sup&gt;c&lt;/sup&gt;</td>
<td>6.00&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.01</td>
</tr>
<tr>
<td>Overall acceptability</td>
<td>7.89&lt;sup&gt;a&lt;/sup&gt;</td>
<td>6.72&lt;sup&gt;b&lt;/sup&gt;</td>
<td>4.50&lt;sup&gt;d&lt;/sup&gt;</td>
<td>5.60&lt;sup&gt;c&lt;/sup&gt;</td>
<td>2.02</td>
</tr>
</tbody>
</table>

abcd: Means on the same row with different superscripts are statistically significant (P < 0.05).

Sensory scores were obtained on a 9-point Hedonic scale where 1 extremely dislike and 9 = extremely like.

4. Conclusion

Fresh beef has high energy and unique biological and chemical properties. However, its nutrients composition represents an optimum medium for microbial growth such that it undergoes deterioration progressively from slaughter until consumption. There is the need therefore for extending the shelf-life of fresh beef cuts, one of the means of extending the shelf-life of beef is by using organic acids. In this study three dietary organic acids were tested viz- citric acid, acetic acid and ascorbic acid while cold preservation (freezing) was used as control since electricity current was supplied throughout the experimental period. It was observed that most of the meat attributes were better in meat samples preserved with 5% acetic acid with exception of colour and flavour which might have contributed to low acceptability of meat sample preserved with the acid. Since other meat attributes were far better in meat preserved with 5% acetic acid, it is hereby recommended that acetic acid should be used in preserving fresh beef and that lower levels of acetic acid be tested in another study to assess the best percentage level that will confer high acceptability on beef preserved using dietary acetic acids.

References


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